

Two crystal polymorphs of a flavonoid from *Melicope ellyrana*

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Received 28 February 2001

Accepted 16 August 2001

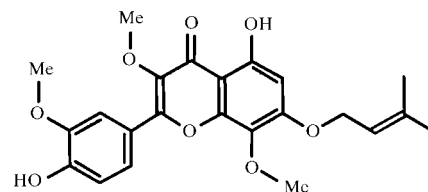
The crystal structure determinations of two crystalline components of the hexane extract of the fruit of the indigenous Australian tree *Melicope ellyrana* have shown them to be polymorphs of the same compound, namely the flavonoid 4',5-dihydroxy-3,3',8-trimethoxy-7-(3-methylbut-2-enyloxy)flavone [systematic name: 5-hydroxy-2-(4-hydroxy-3-methoxyphenyl)-3,8-dimethoxy-7-(3-methylbut-2-enyloxy)-4*H*-1-benzopyran-4-one], C₂₃H₂₄O₈. The two polymorphs, one monoclinic (polymorph *A*) and the other triclinic (polymorph *B*), show significant conformational differences, particularly in the enyloxy side chain, while only one (polymorph *A*) shows intermolecular hydrogen bonding.

Comment

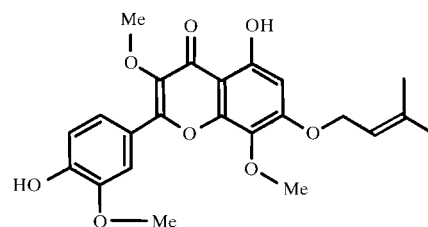
Investigation of the hexane extract of the fruit of the indigenous Australian tree *Melicope ellyrana* (formerly *Euodia ellyrana* F. Meull) (Hartley, 1981) resulted in the isolation of three flavonoids, namely 4',5-dihydroxy-3,3',8-trimethoxy-7-(3-methylbut-2-enyloxy)flavone, (I), pachypodol (4',5-dihydroxy-3,3',7-trimethoxyflavone), (II), and ternatin (4',5-dihydroxy-3,3',7,8-tetramethoxyflavone), (III). Although compound (I) has previously been isolated from *Boronia coerulescens* (Ashan *et al.*, 1994), and other prenylated flavones similar to (I) have been reported, *e.g.* from *M. triphylla* (Higa *et al.*, 1987) and *M. micrococca* (Nasim, 1999), this represents the first report and crystallographic characterization of (I) from *M. ellyrana*.

Two morphologically different crystalline forms of (I) obtained from *n*-hexane (monoclinic polymorph *A*) and ethyl acetate (triclinic polymorph *B*) extracts allowed, in the first instance (because of the minute quantities of the available sample involved), determination of the identity of both polymorphs as the title compound and, in the second instance, confirmation of the existence of two significantly different molecular conformations (Figs. 1 and 2). Triclinic form *B* has a more compact molecular form than is found for monoclinic form *A* due mainly to the more convoluted enyloxy side chain

(comparative molecular volumes of 506 and 526 Å³, respectively). This is reflected particularly in the C7—O7—C71—C72 and O7—C71—C72—C73 torsion angles of 81.9 (4) and 107.4 (5)°, respectively, in polymorph *B*, and −174.6 (3) and 143.5 (4)° in polymorph *A*. In addition, the benzene ring substituent at C2, although essentially coplanar with the



(I), polymorph *A*



(I), polymorph *B*

parent ring, is rotated about the C2—C12 bond vector such that the *meta*-substituted methoxy groups in the two polymorphs are *ca* 180° apart. Intramolecular C—H···O(methoxy) and C—H···O(ether) interactions are significant in maintaining the coplanarity of the two ring systems [C22—H22···O3 2.888 (5) Å and C62—H62···O1 2.627 (4) Å for polymorph *A*; C22—H22···O1 2.664 (5) Å and C62—H62···O3 2.858 (5) Å for polymorph *B*]. Relatively minor conformational differences are found in the methoxy substituents at C3, C8 and C52, while in each, there are intramolecular hydrogen-bonding interactions involving the hydroxyl group at C5 and the adjacent ketone O atom at C4 [O5—H5···O4: 2.600 (4) Å in form *A* and 2.586 (4) Å in form *B*], and the hydroxyl group at C42 and the adjacent methoxy O atom at C52 [O42—H42···O52: 2.692 (4) Å in form *A* and 2.657 (4) Å in form *B*].

A major difference between polymorphs *A* and *B* lies in the packing; in *A*, a single intermolecular hydrogen-bond inter-

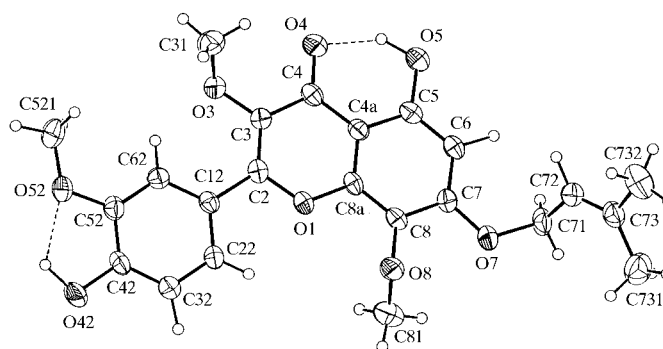


Figure 1

The molecular configuration and atom-numbering scheme for monoclinic polymorph *A*. Atoms are shown as 40% probability ellipsoids.

action is found between the H atom of the hydroxyl group at C42 and a ketone O atom of a glide-related neighbour [O42...H42...O4ⁱ 2.784 (4) Å; symmetry code: (i) $\frac{1}{2} + x, \frac{1}{2} - y, \frac{1}{2} + z$]. This completes a three-centred hydrogen-bond association about H42, with a relatively long H42...O4ⁱ separation of 1.87 Å.

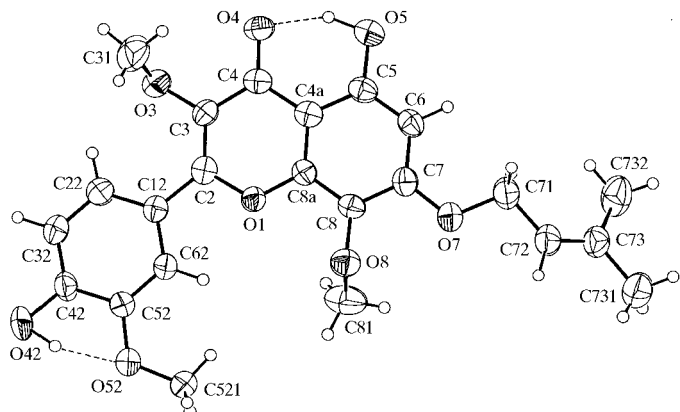


Figure 2
The molecular configuration and atom-numbering scheme for triclinic polymorph B. Atoms are shown as 40% probability ellipsoids.

Experimental

Fresh fruit from *Melicope ellyrana* (1.5 kg) was extracted with *n*-hexane (4.5 l) for 72 h at room temperature and the extract concentrated by rotary evaporation. The extract (20 g) was subjected to gel filtration chromatographic (GFC) separation using methanol (monitored at $\lambda = 254$ nm), the eluent yielding 23 separate fractions. Two of these fractions gave crystals of the title compound in morphologically different and visually identifiable forms, one monoclinic (polymorph A) and the other triclinic (polymorph B).

Compound (I), polymorph A

Crystal data

$C_{23}H_{24}O_8$
 $M_r = 428.42$
Monoclinic, $P2_1/n$
 $a = 14.238$ (4) Å
 $b = 11.053$ (5) Å
 $c = 14.426$ (5) Å
 $\beta = 111.979$ (18)°
 $V = 2105.3$ (13) Å³
 $Z = 4$

$D_x = 1.352$ Mg m⁻³
Mo $K\alpha$ radiation
Cell parameters from 25 reflections
 $\theta = 20.1$ – 27.2 °
 $\mu = 0.10$ mm⁻¹
 $T = 293$ (2) K
Prism, yellow
 $0.40 \times 0.30 \times 0.30$ mm

Data collection

Rigaku AFC-7R diffractometer
 ω - 2θ scans
3865 measured reflections
3709 independent reflections
1532 reflections with $I > 2\sigma(I)$
 $R_{int} = 0.026$
 $\theta_{max} = 25.0$ °

$h = 0 \rightarrow 16$
 $k = 0 \rightarrow 13$
 $l = -17 \rightarrow 15$
3 standard reflections
every 150 reflections
intensity decay: 0.3%

Refinement

Refinement on F^2
 $R[F^2 > 2\sigma(F^2)] = 0.048$
 $wR(F^2) = 0.180$
 $S = 0.95$
3709 reflections
280 parameters
H-atom parameters constrained

$w = 1/[\sigma^2(F_o^2) + (0.0873P)^2 + 0.3178P]$
where $P = (F_o^2 + 2F_c^2)/3$
 $(\Delta/\sigma)_{max} < 0.001$
 $\Delta\rho_{max} = 0.22$ e Å⁻³
 $\Delta\rho_{min} = -0.27$ e Å⁻³

Compound (I), polymorph B

Crystal data

$C_{23}H_{24}O_8$
 $M_r = 428.44$
Triclinic, $P\bar{1}$
 $a = 12.291$ (5) Å
 $b = 13.075$ (5) Å
 $c = 7.099$ (3) Å
 $\alpha = 93.85$ (5)°
 $\beta = 104.33$ (4)°
 $\gamma = 111.52$ (3)°
 $V = 1012.3$ (7) Å³

$Z = 2$
 $D_x = 1.405$ Mg m⁻³
Mo $K\alpha$ radiation
Cell parameters from 17 reflections
 $\theta = 20.1$ – 25.6 °
 $\mu = 0.11$ mm⁻¹
 $T = 293$ (2) K
Prism, yellow
 $0.45 \times 0.10 \times 0.10$ mm

Data collection

Rigaku AFC-7R diffractometer
 ω - 2θ scans
3754 measured reflections
3571 independent reflections
1609 reflections with $I > 2\sigma(I)$
 $R_{int} = 0.097$
 $\theta_{max} = 25.0$ °

$h = 0 \rightarrow 14$
 $k = -15 \rightarrow 14$
 $l = -8 \rightarrow 8$
3 standard reflections
every 150 reflections
intensity decay: 3.5%

Refinement

Refinement on F^2
 $R[F^2 > 2\sigma(F^2)] = 0.055$
 $wR(F^2) = 0.199$
 $S = 1.00$
3571 reflections
280 parameters

H-atom parameters constrained
 $w = 1/[\sigma^2(F_o^2) + (0.1P)^2]$
where $P = (F_o^2 + 2F_c^2)/3$
 $(\Delta/\sigma)_{max} < 0.001$
 $\Delta\rho_{max} = 0.32$ e Å⁻³
 $\Delta\rho_{min} = -0.28$ e Å⁻³

With the exception of those H atoms involved in hydrogen-bonding interactions, H atoms were included at calculated positions and were constrained in the refinement (C–H = 0.94–0.96 Å).

For both compounds, data collection: *MSC/AFC Diffractometer Control Software* (Molecular Structure Corporation, 1999a); cell refinement: *MSC/AFC Diffractometer Control Software*; data reduction: *TEXSAN* (Molecular Structure Corporation, 1999b); program(s) used to solve structure: *SHELXS97* (Sheldrick, 1997); program(s) used to refine structure: *SHELXL97* (Sheldrick, 1997); molecular graphics: *PLATON* (Spek, 1999); software used to prepare material for publication: *TEXSAN*.

The authors acknowledge financial support from The Centre for Instrumental and Developmental Chemistry (Queensland University of Technology) and the Australian Research Council. Dr Peter Healy is thanked for collection of the X-ray diffraction data.

Supplementary data for this paper are available from the IUCr electronic archives (Reference: TA1324). Services for accessing these data are described at the back of the journal.

References

- Ashan, M., Armstrong, J. A., Gibbons, S., Gray, A. I. & Waterman, P. G. (1994). *Phytochemistry*, **37**, 259–266.
Hartley, T. G. (1981). *Gardens Bull. (Singapore)*, **34**, 91–131.
Higa, M., Miyagi, Y., Yogi, S. & Hokama, K. (1987). *Yakugaki Zasshi*, **107**, 954–958.
Molecular Structure Corporation (1999a). *MSC/AFC Diffractometer Control Software*. MSC, 9009 New Trails Drive, The Woodlands, TX 77381, USA.
Molecular Structure Corporation (1999b). *TEXSAN for Windows*. Version 1.06. MSC, 9009 New Trails Drive, The Woodlands, TX 77381, USA.
Nasim, S. (1999). *Phytochemistry*, **50**, 1249–1253.
Sheldrick, G. M. (1997). *SHELXL97* and *SHELXS97*. University of Göttingen, Germany.
Spek, A. L. (1999). *PLATON for Windows*. Version of September 1999. University of Utrecht, The Netherlands.